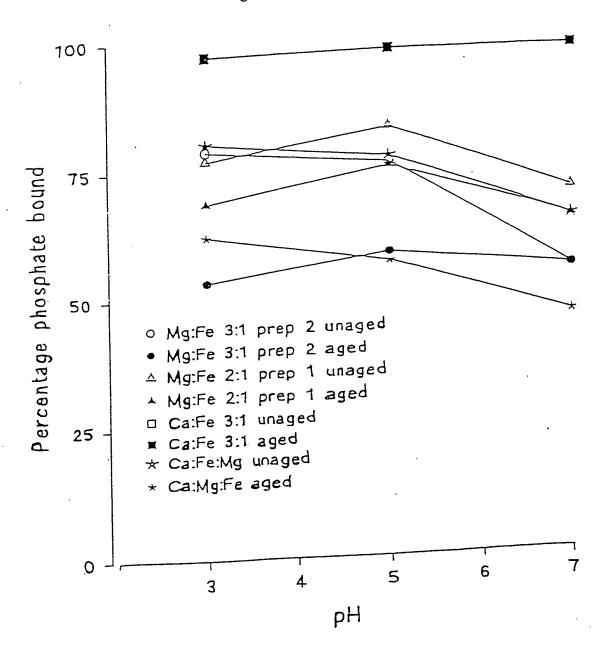
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Figure 1:

Effect of pH and ageing on percentage phosphate binding of mixed metal compounds

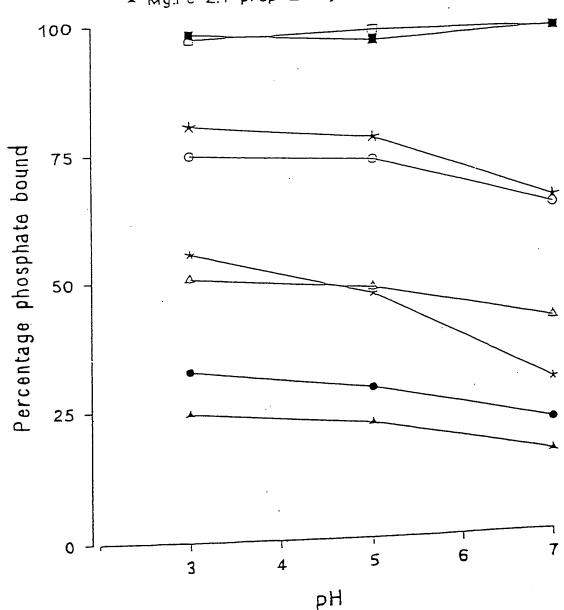


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Figure 2:

Effect of pH and drying on percentage phosphate binding of mixed metal compounds



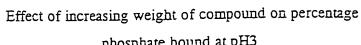


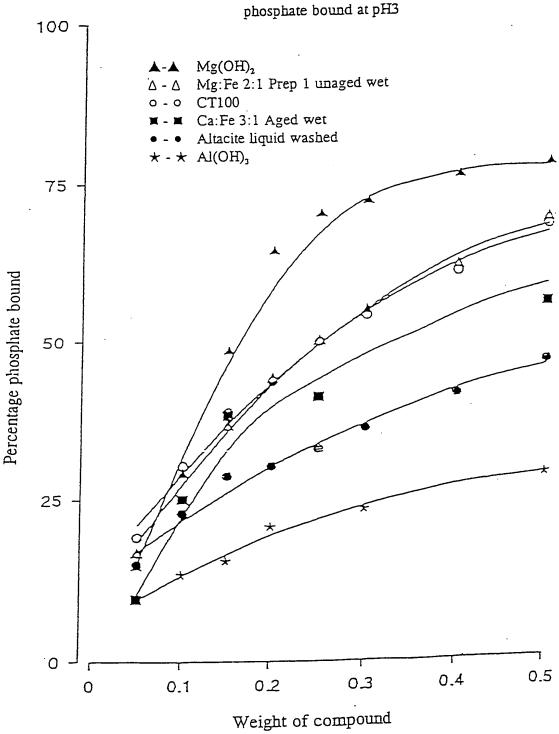
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Figure 3

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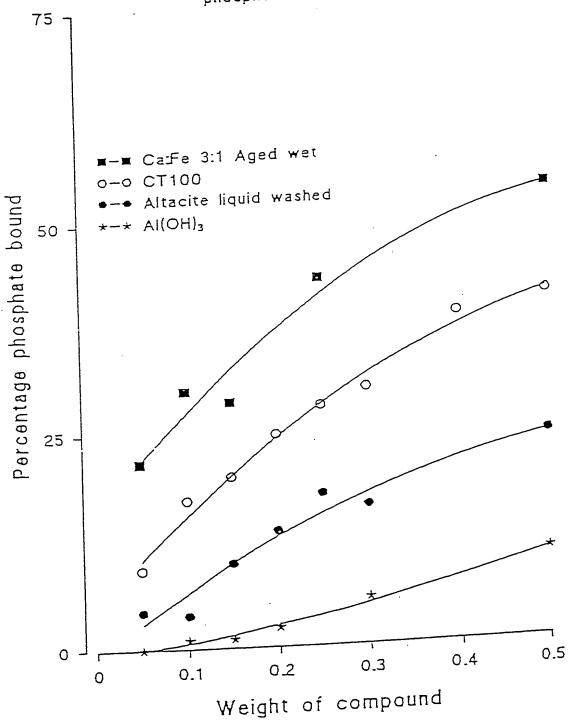


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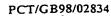
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Figure 4

'Effect of increasing weight of compound on percentage phosphate bound at pH7



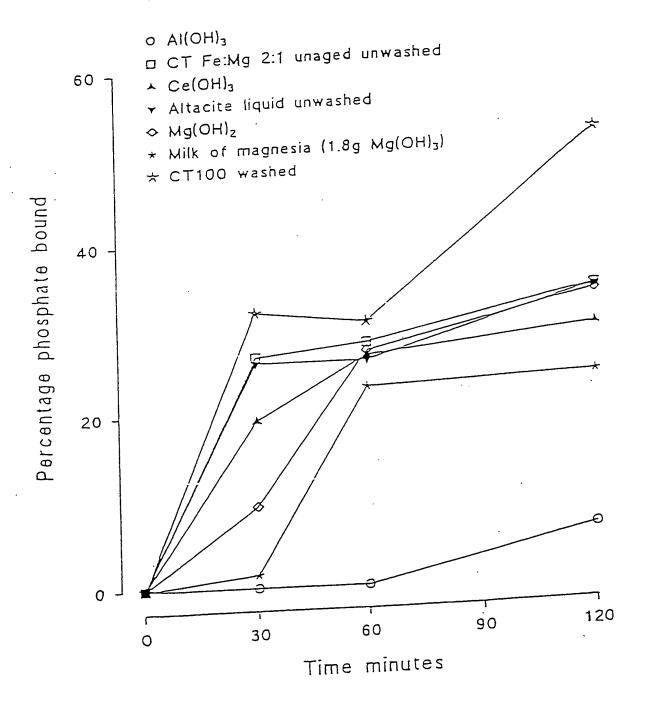
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Figure 5:

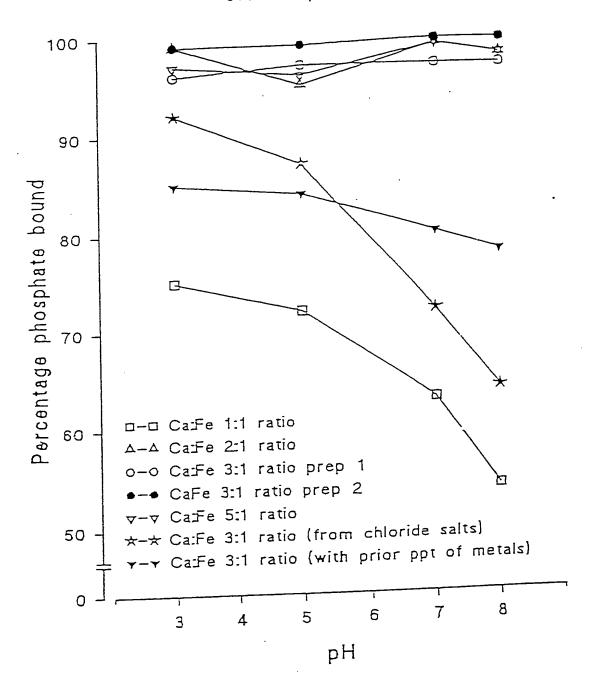
Time course of phosphate binding in food



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Figure 6:

Phosphate binding by the calcium ferric iron preparations over the pH range 3-8



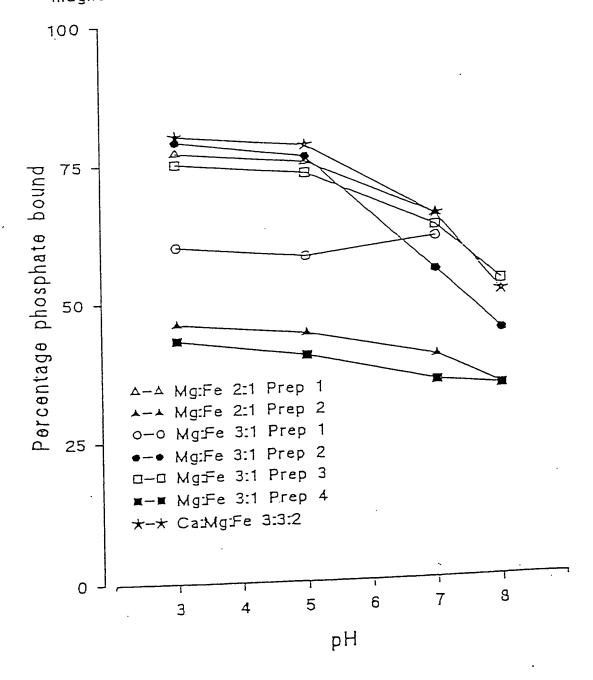
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Figure 7:

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Phosphate binding by the magnesium ferric iron and calcium magnesium ferric iron preparations over the pH range 3-8

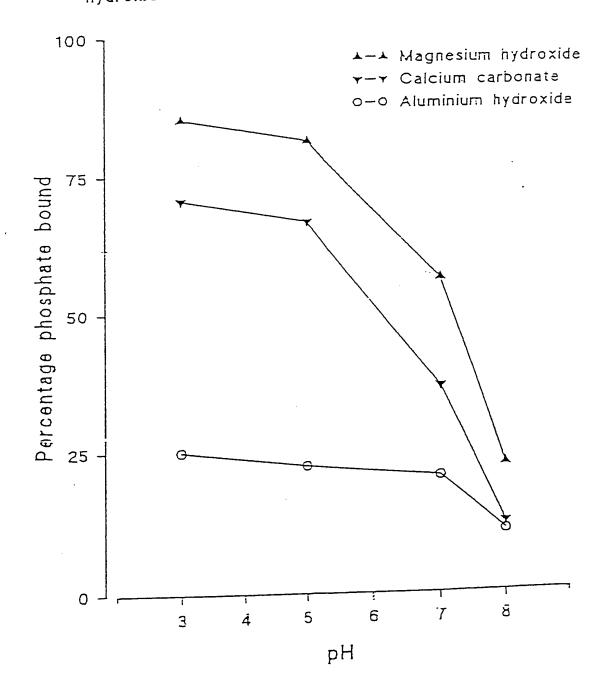


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Figure 8:

Phosphate binding by aluminium hydroxide, magnesium hydroxide and calcium carbonate over the pH range 3-8



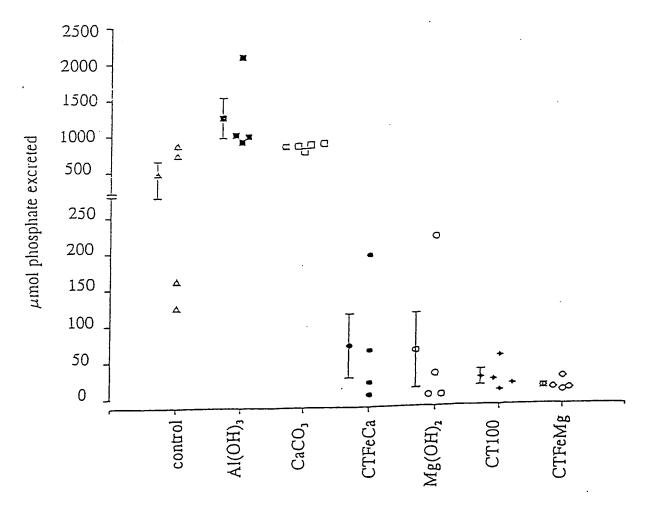
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Figure 9:

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Individual and mean $(\pm 1SEM)$ urinary phosphate excretion for control rats and those treated with phosphate binding compounds.

Individual values of urinary phosphate excretion (μ mol/24 hours) were plotted for controls (Δ) and animals treated with AI(OH)₃ (■), CaCO₃ (□), CTFeCa(\bullet), Mg(OH)₂(\circ), CT100 (\bullet) and CTFeMg (\diamondsuit) . Mean (+ SEM) for each group are presented by points with error bars. *p < 0.05 compared to AI(OH)₃ treated animal groups.

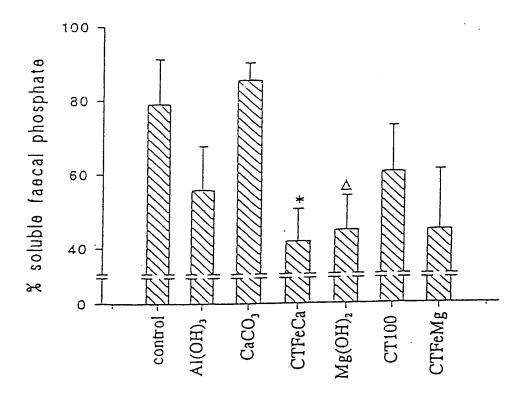


Treatment

Figure 10

Mean (+1SEM) soluble faecal phosphate (g-1 dry weight as a percentage of total soluble and unsoluble) faecal phosphate (g-1 dry weight) for control rats and those treated with phosphate binding compounds.

*p<0.05 compared to control and CaCO3 treated animals Δp <0.05 compared to CaCO3 treated animals



Treatment